


Extracellular DNA in Environmental Samples: Occurrence, Extraction, Quantification, and Impact on Microbial Biodiversity Assessment

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ABSTRACT Environmental DNA, i.e., DNA extracted directly from environmental samples, has been used to understand microbial communities in the environment and to monitor contemporary biodiversity in the conservation context. Environmental DNA often contains both intracellular DNA (iDNA) and extracellular DNA (eDNA). eDNA can persist in the environment and complicate environmental DNA sequencing-based analyses of microbial communities and biodiversity. Although several studies acknowledged the impact of eDNA on DNA-based profiling of environmental communities, eDNA is still being neglected or ignored in most studies dealing with environmental samples. In this article, we summarize key findings on eDNA in environmental samples and discuss the methods used to extract and quantify eDNA as well as the importance of eDNA on the interpretation of experimental results. We then suggested several factors to consider when designing experiments and analyzing data to negate or determine the contribution of eDNA to environmental DNA-based community analyses. This field of research will be driven forward by (i) carefully designing environmental DNA extraction pipelines by taking into consideration technical details in methods for eDNA extraction/removal and membrane-based filtration and concentration; (ii) quantifying eDNA in extracted environmental DNA using multiple methods, including qPCR and fluorescent DNA binding dyes; (iii) carefully interpreting the effect of eDNA on DNA-based community analyses at different taxonomic levels; and (iv) when possible, removing eDNA from environmental samples for DNA-based community analyses.

KEYWORDS extracellular DNA, environmental DNA, eDNA, microbial biodiversity

INTRODUCTION

In recent years, analysis of environmental DNA, i.e., DNA directly extracted from environmental samples, has been applied to understand microbial communities in the environments and to monitor contemporary biodiversity in the conservation context. DNA from an environmental sample often contains both intracellular DNA (iDNA), which represents DNA contained within intact cells, and extracellular DNA (eDNA), which represents DNA outside an organism (1–4). DNA can be released into the environment from organisms through cell lysis and is mediated by predation or phage infection and the release of membrane vesicles (5, 6). Studies have shown that eDNA can persist and accumulate in various environments, such as soil (e.g., relic DNA and adsorbed DNA [ads_DNA] (2, 7)), and aquatic systems (e.g., dissolved DNA [dis_DNA] (8)), although the abundance of eDNA might vary in different environmental systems (7).

eDNA can serve as a source of nutrients to support microbial growth, as a source of free genetic material for iDNA repair and the acquisition of favorable functions, and as a structural component in biofilms (5). In environmental DNA-based biodiversity monitoring through DNA sequencing, eDNA may complicate the determination of the

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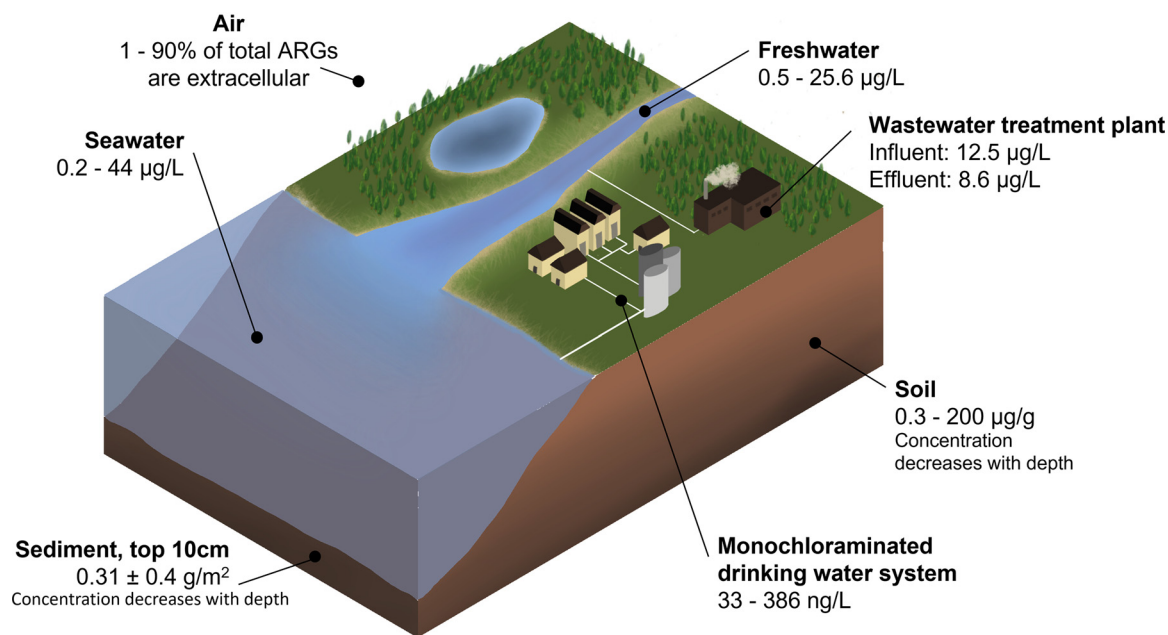


FIG 1 eDNA is ubiquitous and can be found in natural ecosystems (sediments [1, 3, 13 to 18], soil [1, 9–12], air [24], freshwater, and seawater [1, 3, 19, 21]) and engineered ecosystems (wastewater treatment plant [22, 23] and drinking water distribution systems [20]).

structure and potential function of the community inhabiting the sampled environment. As eDNA in environmental samples is a mixture of genomic DNA derived from cell lysis along with plasmids etc., general bioinformatics pipelines cannot effectively distinguish between eDNA and iDNA, highlighting the need to mitigate the effect of eDNA in the design of experiments. Although several studies acknowledged the impact of eDNA on DNA-based profiling of environmental communities, eDNA is still being neglected or ignored in most studies dealing with environmental samples. Here, we summarized key findings on eDNA in environmental samples and discussed the methods used to extract and quantify eDNA as well as the importance of eDNA on the interpretation of experimental results. We then suggested several factors to consider when designing experiments and analyzing data to negate/determine the contribution of eDNA to environmental DNA-based community analyses.

OCURRENCE OF eDNA IN ENVIRONMENTAL SAMPLES

eDNA is ubiquitous. It has been found in natural systems, such as soil, sediments, water, and air as well as in engineered systems, such as wastewater treatment plants, drinking water, and anaerobic digesters (Fig. 1). In soil samples, eDNA concentrations range from 0.3 to 200 $\mu\text{g/g}$ of soil (9) and vary with the soil type and depth from the surface. For instance, podzol was found to contain 2 μg eDNA/g of soil, luvisol contained 0.08 μg eDNA/g of soil (10), fine loamy soil contained 60 μg eDNA/g of soil (11), and Cambic Umbrisol contained 6.07 μg eDNA/g of soil (12) in the topsoil layer. eDNA concentrations have been observed to decrease with an increase in the depth of soil. In the subsoil layer, 0.4 ng eDNA/g of soil was found in luvisol, 5.2 ng eDNA/g of soil in podzol, and 10.8 μg eDNA/g of soil in fine loamy soil. Sediment samples exhibit a similar trend. A study on sediments from Lake Towuti in Indonesia showed that the highest concentration of eDNA was found in the top 5 cm layer of sediment at both shallow and deep sites (0.5 to 0.6 μg eDNA/g wet sediment) and decreased to below detection limits at a depth of 30 to 35 cm (13). Various amounts of eDNA have been successfully extracted from the sediments, which have been summarized recently by Torti et al. (3). Studies on sediments from the Baltic Sea, Barents Sea, South Pacific Gyre (14), Adriatic

Sea, Mediterranean Sea, and South Pacific Ocean (15) have reported large amounts of eDNA, which are 6- to 68-fold higher than the amount of iDNA extracted from the same samples. Mao et al. (16) also reported higher eDNA than iDNA extracted from river sediments. A recent study on sediments from Aarhus Bay showed that only 40% of the total DNA was extracellular (17). Globally, the top 10 cm of deep-sea sediment is the largest reserve of eDNA, accounting for ~90% of the total DNA (18).

In the marine environment, eDNA in water columns can range between 0.2 to 44 μg eDNA/L of water depending on the location of sampling (estuarine versus coastal versus offshore) while freshwater ecosystems can harbor 0.5 to 25.6 μg eDNA/L of water (3). More recently, 10.3 μg eDNA/L of water was found in the hypersaline lake environment (19). Monochloraminated drinking water systems were also shown to have a low quantity (33 to 386 ng/L) but a significant proportion (~50%) of eDNA in total DNA (20). Another study detected 0.12 to 2.5 μg eDNA/L in the Tama River water (21). Zhang et al. (22) showed the presence of extracellular antibiotic resistance genes (eARGs) in wastewater treatment plants and the eARGs proportion increased along with the treatment plant. eARGs were also seen to decay slowly in the treated water indicating their persistence. Calderon et al. (23) estimated influent and effluent wastewater to contain 12.5 μg and 8.6 μg eDNA/L, respectively. Activated sludge was also found to contain substantial amount of eDNA (12.3 μg /L) (23). Although reduced amounts of eDNA were found in the water column compared to the soil and sediment, eDNA still formed a major proportion of the total DNA extracted from water samples.

The presence of eDNA in the discussed system has been known for 3 decades with the earliest reports arising in the mid-1980s. A new study showed that, apart from these conventionally studied systems, air can also contain eDNA. The study focused on the presence of eARGs adsorbed to the PM_{2.5} particles and showed that the ARGs in the iDNA fraction were distinct from the eDNA fraction (24). Overall, eDNA contributes significantly to total DNA extracted from environmental samples, and further studies are needed to understand its implication, fate, and biological significance in different environments.

PERSISTENCE OF eDNA IN ENVIRONMENTAL SAMPLES

The interaction of eDNA with extracellular nucleases is one of the most important factors influencing its persistence in the environment. Environmental temperature affects nuclease activity and, hence, eDNA persistence. DNA molecules are known to adsorb onto particles, such as sands, clays, and minerals (25), that can protect them from nucleases (26). The length of eDNA inversely correlates to its adsorption potential and, hence, smaller eDNA fragments are more persistent than longer fragments (27). Other factors, such as low temperature and anoxic conditions, also help in the preservation of eDNA (28, 29). Another major factor influencing eDNA persistence is its metabolism and uptake by microorganisms. Certain groups of bacteria (such as *Lutibacter*, *Shewanella*, "*Candidatus* Izemoplasma", and *Fusibacteraceae*) in deep-sea sediments were shown to be capable of utilizing eDNA as a carbon source. In soil samples, bacteria belonging to the genera *Arthrobacter* and *Nocardioides* may also play a role in the degradation of eDNA (30, 31).

Degradation of eDNA in various environments has been previously summarized by Nielsen et al. (1) and Pietramellara et al. (9) and can vary between different environmental matrices (Fig. 2). More recently, Sirois et al. (32) reported that the persistence of eDNA in soil is positively correlated with its organic matter content while higher moisture and temperature enhanced eDNA degradation. Although the eDNA standard (469 bp *htrA* gene fragment) was observed to decline rapidly upon its introduction into soil microcosms (~99% reduction in 7 days), the eDNA standard could be detected for up to 80 days. A similar phenomenon was reported by Kunadiya et al. (33), where degradation of DNA from *Phytophthora cinnamomi* was accelerated in moist soils compared to dry soils. eDNA in dry soils could be easily detected for up to 90 days using qPCR. Gordon et al. (34) also reported the successful detection of *Phaeocollybia* eDNA up to 60 days in soil samples. Plasmids incubated in soil microcosms are also known to persist for at least

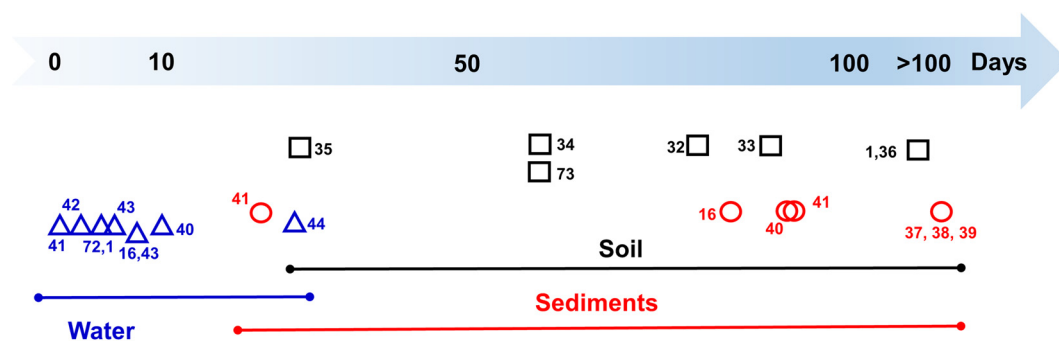


FIG 2 Persistence of eDNA in water (Δ), soil (\square) and sediments (\circ). eDNA can persist in water for 10 h to 28 days (16, 40–44, 72), while eDNA in soil (1, 32–36, 73) and sediments (16, 37–41) can persist for a minimum of 24 to 28 days up to thousands of years. Numbers next to the symbols are reference numbers for the sources of literature. The reported results are observations from different studies, considering all factors, including environmental matrices, that may influence eDNA persistence in respective experimental systems.

28 days (35). eDNA introduced as cell lysate into soils was seen to persist for 24 weeks (36).

DNA is effectively preserved in the marine sediments as evidenced by retrieval of eDNA as old as 217,000 years (37, 38) and the successful amplification of 16S rRNA genes from eDNA extracted from sediment samples that were 10,000 years old (39). Using microcosms, Mao et al. (16) reported that eDNA degraded faster in Haihe river water (~100% within 7 days) than in the Haihe river sediments (detectable after 12 weeks). Deere et al. (40), also observed eDNA to persist in the water column for 10 days, whereas it could still be detected in sediments for up to 13 weeks. Dell'Anno et al. (41) estimated the turnover time for DNA in sediments to be longer than water (29 to 93 days in sediments versus 10 h in seawater). In forest pond sediment-water microcosms, England et al. (42) reported that viral eDNA from *Baculovirus* could be detected for 24 h, whereas the eDNA could persist for days in laboratory microcosms. Saito et al. (43) reported that eDNA was not degraded in purified water over the experimental period of 7 days, whereas it persisted in pond water for 5 days and could be detected in seawater after 7 days. Similar to Mao et al. (16), Saito et al. (43) also observed rapid initial decay of eDNA in environmental water samples, which were attributed to the microbial activity and action of extracellular enzymes. However, a study by Bukh et al. (44) reported eDNA to persist for at least 28 days in hot tap water. The rapid decay of eDNA in environmental waters compared to engineered water systems could also be attributed to dissolved organic matter, e.g., humic substances, organic acids, and carbohydrates. This organic matter can be photosensitized by sunlight, resulting in the formation of reactive oxygen species (ROS) and hydroxyl radicals which facilitate the decay of DNA (45, 46). Overall, DNA seems to persist longer in sediments and soils compared to water samples. This might be attributed to faster eDNA decay in water because of reduced protection from extracellular nucleases, increased microbial activity (43, 47), a high chance of ROS-induced DNA damage and higher temperatures (48, 49), or a reduced detection efficiency due to lower DNA concentrations in water environments and DNA binding to organic matter (50).

Due to its persistence, eDNA serves as a possible reservoir of antibiotic resistance genes (ARGs) for horizontal gene transfer via transformation. Uptake of eDNA by competent bacterial cells can result in the dissemination of these genes on a large scale. Mao et al. (16) demonstrated the uptake of the kanamycin resistance gene by an indigenous sediment bacterium under selective pressure. Recent research has suggested that disinfection by-products from chlorine and chloramine can lead to an increase in uptake of exogenous DNA (51). Disinfection using chlorine, chloramine (52), and solar irradiation (53) have also been seen to enhance the transformation process through upregulation in DNA uptake and repair proteins caused by the ROS stress response.

Biofilms in estuarine systems have been shown to accumulate ARGs in eDNA (54) facilitating the horizontal gene transfer process (55). However, most studies are performed under ideal laboratory conditions using model organisms and plasmids. Further studies are required to determine the real contribution of eDNA in the environment as a pool for horizontal gene transfer.

EXTRACTION AND QUANTIFICATION OF eDNA IN ENVIRONMENTAL SAMPLES

Table 1 summarizes previous studies on eDNA in environmental samples. The most common method to extract eDNA from solid phases, such as soil samples is to desorb the adsorbed eDNA using an alkaline sodium phosphate buffer (NaP) (56) (Fig. 3). The excess phosphate in NaP competes with eDNA for binding sites on clay particles, enabling the desorption and recovery of eDNA from the samples. This method has been widely used with various modifications over the last 3 decades (11, 13–15, 57). A remarkable improvement in the yield of eDNA (~4 to 10-fold) was observed when the samples were pretreated with proteinase K (21). Another method for eDNA extraction from solid phases is to dissolve eDNA in Tris EDTA (TE) buffer from the environmental samples (10, 17, 58) (Fig. 3). Treatment with NaP/TE or proteinase K does not cause cell lysis, thereby preventing contamination of eDNA with iDNA. The obtained crude extracts of eDNA can be purified using conventional DNA extraction techniques such as the cetyltrimethylammonium bromide (CTAB) method, ethanol precipitation, or chromatography. These methods have mainly been applied to soil, sediments, and sludge samples due to practical feasibility as the particles can be resuspended and washed in the extraction buffer.

To extract eDNA from liquid environmental samples, we usually need to concentrate it first (Fig. 3). For example, eDNA in water samples can be concentrated through adsorption by the nucleic acid adsorption particles (NAAPs; silica coated with aluminum hydroxide) followed by elution and precipitation (59). This method has mainly been used to extract extracellular antibiotic-resistant genes (eARG) from different water systems as the NAAPs favor adsorption of short-length linear DNA (59). Another method used a hollow fiber membrane (MWCO of 30 kDa and a surface area of 25 m²) and silica adsorption to concentrate and extract eDNA, from Tama River water (21). Compared with the NAAPs method, this method gives a higher recovery by allowing the concentration of all DNA to be above 50 bp. Calderon et al. (23) used anion exchange chromatography for eDNA extraction from wastewater samples and found that plasmids and transposable elements were highly enriched in the extracted eDNA. Yuan et al. (60) reported the use of a prefiltration technique combined with magnetic beads to isolate eDNA from water and sludge samples. Using the optimized method, large quantities of free and adsorbed eDNA from small volumes of wastewater and activated sludge samples could be extracted.

Different filter membranes have been used to concentrate eDNA from water samples. Glass fiber (GF) and cellulose nitrate (CN) membranes exhibit the highest affinity to eDNA (~30 to 35% of eDNA retained) (61), followed by polyvinylidene difluoride (PVDF), mixed cellulose esters (MCE), polyether sulfonate (PES) and polycarbonate (PC) membranes (62). Hence, GF or CN filter membranes are recommended for extracting eDNA from the filter retentate, while PC and PES membranes are preferable for separating eDNA from cells and recovering eDNA from the filtrate (Fig. 3). In addition to the adsorption by membrane filters, the retention of eDNA on filters may also be attributed to other components in the samples. For example, using PC membrane for the filtration of stormwater, Liang et al. (62) found that 7 to 13% of their qPCR signals from the filter retentate were derived from the spiked plasmid DNA (eDNA proxy) even though they observed only ~2% eDNA retention in optimization experiments with PC membranes. The increased detection of the spiked plasmid DNA in the retentate can be attributed to the adsorption of eDNA on sand and clay particles retained on the filter membranes (62).

Extracted eDNA is usually quantified by using fluorometric DNA dyes, such as Picogreen (10, 58, 63), Hoechst (8, 11), SYBR green (15), Qubit dsDNA assay kits (2, 20, 21), or DAPI (61). DNA quantification using fluorometric dyes is more specific and

TABLE 1 Summary of previous studies dealing with eDNA in environmental samples^a

Sample	Filter	Starting material	Method of extraction/ treatment	Quantification	eDNA yield	Recovery/ removal efficiency	Impact on 16S/ 18S rRNA genes detection	Ref
Soil	CA	2 g	TE Buffer dissolution	Picogreen	Podzol: 1949.5 ng/g; Luvisol: 80.8 ng/g (top layer) ~18.9 μg/g (top layer), 10-60% eDNA over all horizons	-	-	(10)
	-	10 g	NaP wash	Hoechst 33258		-	Yes - DGGE	(11)
	-	0.5 g	NaP wash	Hoechst 33258	Total eDNA: ~8.2 μg/g soil Free eDNA: ~1.4 μg/g soil Bound eDNA: ~6.9 μg/g soil (Top layer)	-	Yes - DDGE	(57)
	-	0.25 g	PMA/EMA	Picogreen/Nanodrop	Total DNA reduced by 14.9% (PMA, 0.56 μg/g soil) and 16.4% (EMA, 0.74 μg/g soil)	-	No - DGGE	(63)
	-	0.03 g	PMA	qPCR	~41 % of amplifiable soil DNA pool	100% recovery of spiked eDNA	Yes	(2)
	-	0.03 g	PMA	qPCR	30-97% of amplifiable soil DNA pool	-	Yes	(67)
	-	0.03 g	PMA	Qubit ds DNA	Not reported. Calculated: ~1.4 μg/g soil (top layer) in control samples.	-	No	(65)
Sediment	-	100 g	NaP wash	Spectroscopy	1 μg/g sediment	-	-	(56)
	PC	Buffer/Sediment ratio ~2.5 (v/wt)	TE Buffer dissolution - Enzymatic treatment	Diaminobenzoic acid	7.8 - 22 μg/g sediment (top layer)	Higher than alkaline method	-	(74)
	Anopore	2.5 g	NaP wash	SYBR green I	6.7 - 24.3 μg/g sediment	34-60% using internal DNA standard	No	(15)
	CA	Buffer/Sediment ratio ~2 (v/wt), 2-2.5 g	NaP wash	Qubit ds DNA	699 - 889.6 ng/g sediment	13-91%	Yes - RFLP	(14)
	-	0.2 g	Carbonate dissolution followed by alkaline treatment and TE buffer dissolution	Picogreen, qPCR	10-83% of total DNA based on qPCR, 49-61% based on fluorescence spectroscopy	-	-	(58)
	-	1 g	Carbonate dissolution - alkaline treatment - TE buffer dissolution	Qubit ds DNA	0.6 to 0.5 μg/g sediment (top layer)	-	Yes	(13)
	-	0.1 cm ³ of fresh wet sediment	PMA	qPCR	Pacific 36-50%; Arctic 28%	PMA-treatment removed 73- 98% of eDNA	Minimal difference	(64)
	-	0.2 g	Carbonate dissolution - alkaline treatment -TE buffer dissolution	Picogreen, qPCR	0.03-4.45 μg/cm ³ , 28-58% (Picogreen), 20-57% and 33-69% (qPCR)	-	Yes	(17)

(Continued on next page)

TABLE 1 (Continued)

Sample	Filter	Starting material	Method of extraction/ treatment	Quantification	eDNA yield	Recovery/ removal efficiency	Impact on 16S/ 18S rRNA genes detection	Ref
Water	GF/D, PC	100-1000 mL	0.2 μm prefiltration – filtrate precipitated with ethanol	Hoechst 33258	Dissolved DNA constituted 11-226% of the particulate DNA. Marine environment 0.41- 14.52 μg/L, Freshwater 1.7-7.8 μg/L	85-95%	-	(8)
	PC	100-1000 mL	0.2 μm prefiltration - Filtrate precipitated by CTAB-NaCl	DAPI	2.5-72 μg/L	80-90%	-	(61)
	-	1000 mL	PMA treatment	-	-	-	No	(75)
	PC	100 mL	0.2 μm filtration - eDNA extraction from filter residue	qPCR	7-13% of spiked plasmid recovered from storm water	-	-	(62)
	NAAP GF/F	10 L 15 L	NAAP adsorption DNase treatment	TaqMan qPCR Qubit dsDNA assay	-	95% ~99% eDNA removal	-	(59) (20)
	PES	20 L	Proteinase K + NaP wash	Qubit dsDNA assay, qPCR	Location 1: 85-386 ng/L, Location 2: 33-58 ng/L iDNA: 0.15-7.40 μg/L, Ads_eDNA: 0.06-0.45 μg/L, Dis_eDNA: 0.06-2.37 μg/L	38-63%	-	(21)
Multiple	PVDF	Sediment: 1 g, Water: 200 mL	NaP wash	Spectrophotometer	Haihe River sediment: 96.8 ± 19.8 μg/g, Haihe River water: 2.2 ± 0.8 μg/ mL	37-81%	-	(16)
	PES	250 mL, 0.25 g	DNase treatment	qPCR	Soil: ~22%, Sediment: ~30%, Gut: ~40%, Water: ~48%, range 0 – 83%	98% eDNA removal	Minimal difference	(7)
	PES	1000 mL	0.2 μm Prefiltration - DEAE cellulose chromatographic column	Qubit dsDNA assay	Influent: ~12.5 μg/L, Activated sludge: ~12.3 μg/L, Effluent: ~8.6 μg/L	-	No	(23)

^aCA, cellulose acetate; PC, polycarbonate; GF, glass fiber; PES, polyether sulfonate; PICA, phenol-chloroform-isoamyl alcohol; NaP, sodium phosphate buffer; PMA, propidium monoazide; EMA, ethidium monoazide; CTAB, cetyltrimethylammonium bromide; NAAP, nucleic acid adsorption particles; DEAE, diethylaminoethyl cellulose; DGGE, denaturing gradient gel electrophoresis.

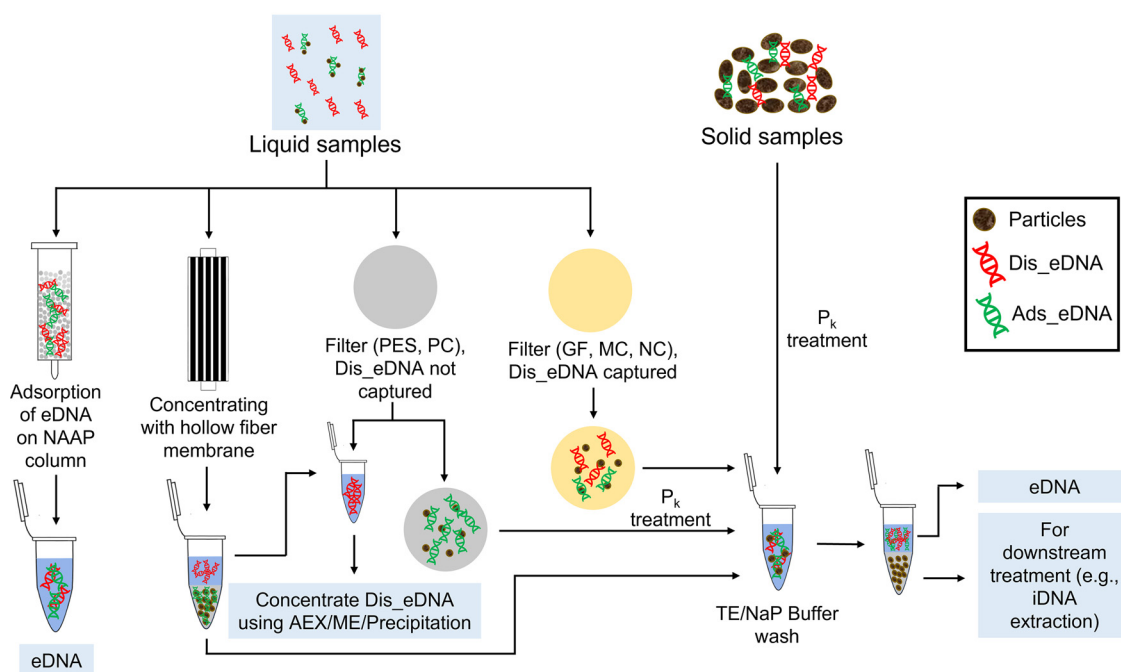


FIG 3 Isolation of eDNA from solid and liquid environmental samples. NaP: sodium phosphate buffer; TE: Tris EDTA buffer; iDNA: intracellular DNA; eDNA: extracellular DNA; Dis_eDNA: dissolved eDNA; Ads_eDNA: adsorbed eDNA; NAAP: nucleic acid adsorption particles; GF: glass fiber membrane; MC: mixed cellulose ester membrane; NC: nitrocellulose membrane; PES: polyether sulfonate membrane; PC: polycarbonate membrane; AEX: anion exchange column; ME: magnetic beads; P_k: proteinase K.

accurate compared to absorption spectroscopy methods which are nonspecific. For instance, freshwater ecosystems usually harbor 0.5 – 25.6 μg eDNA/L of water (3). A study conducted in the Haihe River quantified eDNA using absorption spectroscopy and reported eDNA concentrations of $\sim 2,000$ μg eDNA/L in the water samples (16). This concentration is likely overestimated because of the low specificity of the spectrometric DNA quantification method.

qPCR can also be used to quantify eDNA using specific target genes such as the 16S rRNA or 18S rRNA gene. Interestingly, results from fluorometric assays and qPCR are not always congruent. For example, studies using the alkaline method for eDNA extraction reported a very high fraction of eDNA in total DNA with the eDNA/iDNA ratio of 6 to 68 (14, 15) in sediment samples whereas qPCR suggested only ~ 10 to 80% (average $\sim 40\%$) of the 16S rRNA genes originated from eDNA (7, 58). The discrepancy could be because the use of qPCR for target genes such as the 16S rRNA gene often substantially underestimates eDNA content in environmental samples. For example, although eDNA was abundant in marine sediments, qPCR failed to amplify the 1,500 bp 16S rRNA gene from the eDNA fraction (15). Low integrity of eDNA, the presence of PCR inhibitors, the copy number of the target marker gene, and target amplicon size are factors contributing to the high variability in eDNA quantification by qPCR. For example, in Aarhus Bay sediment samples, qPCR for bacteria and archaea suggested that eDNA as 42 to 51% and 29 to 71% of total DNA, respectively, while fluorescence spectroscopic measurement showed the eDNA fraction to be 49 to 61% (58).

EFFECT OF eDNA ON DNA-BASED COMMUNITY ANALYSES AND ITS MITIGATION

The effect of eDNA on metagenomic studies may vary for different environmental samples. Lennon et al. (7) showed, using a statistical model, that if the pool of eDNA is similar to iDNA, the presence of eDNA would not affect DNA-based biodiversity analyses. Otherwise, eDNA could cause overestimation or underestimation of biodiversity. Hence, the effect of eDNA on DNA-based community analyses depends on factors causing dissimilarity between eDNA and iDNA, for example, decay of eDNA in the

given environmental conditions. Lennon et al. (7) reported that although eDNA accounted for a high fraction of total DNA in sediment, soil, gut, and surface water samples, no significant effect of eDNA on the richness and evenness of the detected microbial communities was observed. Gustavo et al. (64) observed significant differences in 16S rRNA gene copy numbers in 3 out of 12 samples. After removing the eDNA signal from the 3 samples, no significant change was observed for community composition. However, 3 out of the top 16 detected OTUs showed eDNA-based fluctuations across different depths in the sediment samples. Gustave et al. (65) reported a minimal influence of eDNA on community composition in paddy soil microbial fuel cells but also showed significant changes in the relative abundance of certain taxa at the genus level after removing the eDNA signal.

Carini et al. (2) reported a significant difference of eDNA (~40% of total DNA) removal on the soil community detection. The presence of eDNA in soil samples resulted in the overestimation or underestimation of some bacterial and fungal taxa via amplicon analysis. In addition, the spatial (66) and temporal (67) shifts in community structure were found more apparent in analyses excluding eDNA in soil samples and anaerobic digesters (68). Torti et al. (17) also reported OTUs unique to eDNA in sediment samples from the Aarhus Bay. In our recent study, we reported a significant effect of eDNA removal on community analysis at the genus level in drinking water (20).

The effect of eDNA on community characterization is more discernible at the ASV/OTU or genus level. The abundance of a higher taxonomic level (e.g., phylum or class level) is a summation of abundances of lower taxonomic level (i.e., an abundance of a class is a summation of abundances of all orders under it. The abundance of each order is a summation of abundances of the family under it and so on). Thus, increase or decrease in abundance of genera at a higher taxonomic level (e.g., phylum or class level) may not be evident or accurately determined; at lower taxonomic levels, upon eDNA removal, some taxa are under detected while some are over detected. At higher taxonomic ranks, such changes would be confounded and become undetectable.

To mitigate the effect of eDNA on DNA-based community analyses, we can either remove eDNA from the total DNA or inhibit the amplification of eDNA in PCR using propidium monoazide (PMA) for DNA metabarcoding (Fig. 4). eDNA can be extracted using the discussed method (Fig. 3) and the remaining biomass can be used to extract iDNA. Nucleases can also be used to degrade eDNA, while iDNA may remain intact when the treatment conditions are carefully optimized (7, 20, 69). Lennon et al. (7) reported an eDNA removal efficiency of 97 to 99% using DNase on soil samples, while for drinking water samples we reported ~99% eDNA degradation efficiency (20). DNA-intercalating dyes such as PMA can be used to inhibit the PCR amplification of eDNA (2, 63, 64, 67, 70). Treatment of samples with optimized protocols for PMA treatment minimized contamination of eDNA with iDNA-based community analyses. Nocker et al. (71) evaluated PMA for cell lysis of various Gram-positive and Gram-negative bacteria and concluded that it did not lyse intact cells. Similarly, Carini et al. (2) reported that PMA treatment did not cause lysis of exponentially growing cells of *Escherichia coli* and *Saccharomyces cerevisiae*. However, the efficiency of these treatments might vary from sample to sample. In a comparative study, Villarreal et al. (69) reported comparable performance of DNase and PMA treatments in differentiating live/dead cells in drinking water biofilms and suggested DNase treatment as a “more practical alternative” to the PMA-qPCR method. Using PMA, Gustavo et al. (64) reported an eDNA PCR signal removal efficiency of 73 to 98% in sediment samples, while Carini et al. (2) reported that PMA could not remove PCR signals of the spiked DNA in one sample type. Wagner et al. (63) reported no difference in DGGE patterns observed between total and PMA-treated soil samples (~15% reduction in DNA yield after PMA treatment), whereas Agnelli et al. (11) have observed significant differences in DGGE banding of total and extracellular DNA (10 to 60% of total DNA) extracted using the alkaline method. These studies suggest that the removal efficiency of PCR signals from eDNA using the PMA treatment may vary substantially for different environmental samples, depending on the

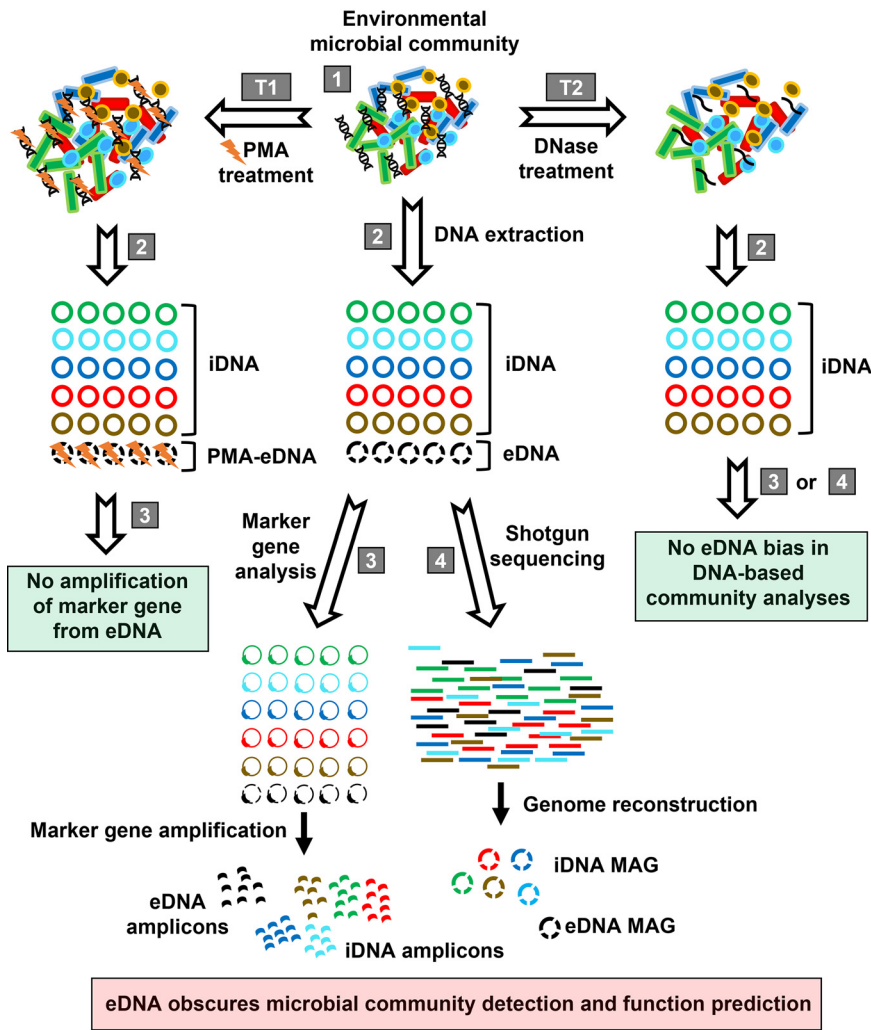


FIG 4 A schematic showing the effects of eDNA on metagenomic studies and ways to mitigate them. In a typical metagenomic study, microbial communities obtained from the environments (1) are subjected to DNA extraction (2) to obtain total DNA containing both iDNA and eDNA. The DNA can be analyzed for the presence of marker genes by amplifying them using PCR followed by sequencing the amplicons (3). The DNA can also be directly sequenced using shotgun sequencing (4) and genomes of sampled organisms (MAGs, metagenome-assembled genomes) can be reconstructed. Effects of eDNA can be mitigated using PMA treatment (T1) or DNase treatment (T2). PMA treatment inhibits PCR amplification of marker genes from eDNA, while DNase treatment removes eDNA from the environmental samples before DNA extraction.

physicochemical properties of the sample matrices and other factors influencing PCR efficiency. Although DNase treatment has been demonstrated effective in mitigating the influence of eDNA on metagenomic studies, it does not apply to environmental samples containing DNase inhibitors that may substantially reduce eDNA removal efficiency.

The choice of approach depends largely on the research objectives and practical considerations, for example, sample types, number of samples, and inhibitors. Extraction of eDNA and iDNA separately can be used if the research questions require the characterization of eDNA. This approach differentiates the two fractions of total DNA, allowing the determination of the true community structure and metabolic potential by analyzing the iDNA fraction. eDNA analysis can be used to determine its composition and if it harbors important genes such as antibiotic resistance genes and mobile genetic elements which can be disseminated to naturally competent bacterial cells via transformation. A major limitation of this method is the recovery efficiency of eDNA from environmental samples.

Due to intrinsic variability in environmental samples, most studies report a large variation in extraction efficiencies ranging from 10 to 90%, which may cause a discrepancy in data interpretation (Table 1). In contrast, DNA removal strategies (DNase/PMA treatment) show a higher efficiency of eDNA removal across studies (73 to 99%). However, these treatments complicate downstream computational analyses. Optimization is needed as the effectiveness of DNase/PMA treatment varies between different sample types. Using these approaches, the contribution of eDNA may be indirectly inferred by examining the difference in total DNA- and iDNA-based community analyses.

CONCLUDING REMARKS

DNA-metabarcoding and metagenomics have revolutionized the way we understand natural and engineered ecosystems. While the methods are evolving, some fundamental issues need to be addressed. Specifically, there is an immediate need for researchers to acknowledge the presence and persistence of eDNA in environmental samples and mitigate its effect on DNA-based community analyses. This field of research will be driven forward by (i) carefully designing environmental DNA extraction pipelines by taking into consideration technical details in methods for eDNA extraction/removal and membrane-based filtration and concentration; (ii) quantifying eDNA in extracted environmental DNA using multiple methods, including qPCR and fluorescent DNA binding dyes; (iii) carefully interpreting the effect of eDNA on DNA-based community analyses at different taxonomic levels; and (iv) when possible, removing eDNA from environmental samples for DNA-based community analyses.

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